

# ANALYSIS OF IRON CONTENT IN HEMOGLOBIN

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**ABSTRACT:**This article is about hemoglobin and its analysis which iron. Hemoglobin is a protein found in red blood cells that plays a crucial role in transporting oxygen from the lungs to the tissues in our body. Hemoglobin, a complex protein found in red blood cells, plays a crucial role in oxygen transport throughout the body

Key words:Hemoglobin, analysis, oxygen-carrying capacity, Spectrophotometry, iron, polycythemia, chromatography.

Hemoglobin is a protein found in red blood cells that plays a crucial role in transporting oxygen from the lungs to the tissues in our body. Hemoglobin, a complex protein found in red blood cells, plays a crucial role in oxygen transport throughout the body. Understanding the structure, function, and clinical significance of hemoglobin is essential for diagnosing and managing various medical conditions related to blood disorders[1]. The core of the hemoglobin molecule contains an iron atom, which is essential for its oxygen-carrying function. Hemoglobin is a vital protein in red blood cells responsible for transporting oxygen from the lungs to the body's tissues and organs. The core of the hemoglobin molecule contains iron, which is essential for its oxygen-carrying capacity. Structurally, hemoglobin consists of four subunits, each containing a heme group with an iron atom at its core[2]. This iron atom is what allows hemoglobin to bind and transport oxygen. The interaction between oxygen and hemoglobin is essential for oxygen uptake in the lungs and its release to the tissues, ensuring proper cellular function. Analyzing the iron content in hemoglobin is crucial for understanding its functionality and diagnosing related disorders like anemia.

Functionally, hemoglobin's ability to bind oxygen is known as oxygen affinity. Factors such as pH, temperature, and the concentration of carbon dioxide can influence this affinity, impacting oxygen delivery to tissues. Understanding these factors is critical in conditions like hypoxia, where oxygen delivery to tissues is compromised[3]. Quantifying the iron content in hemoglobin is an important aspect of medicine and biomedical research, as iron deficiency can lead to health problems such as anemia. Therefore, having accurate and reliable methods to analyze the amount of iron in hemoglobin is crucial. Several methods are employed to assess the iron content in hemoglobin accurately. Spectrophotometry, a common technique, measures the absorption of light by iron atoms, enabling precise determination of iron concentration in hemoglobin samples. This method is effective due to the distinct light absorption properties of iron[4].

There are several methods for analyzing the iron content in hemoglobin. The clinical significance of hemoglobin analysis lies in its diagnostic capabilities. Hemoglobin levels are routinely measured in blood tests to assess oxygen-carrying capacity[4]. Abnormalities in hemoglobin levels can indicate various conditions, such as anemia, polycythemia, or genetic hemoglobin disorders like sickle cell disease or thalassemia[1]. Additionally, hemoglobin analysis is crucial in monitoring patients with chronic diseases such as kidney failure or cancer, where anemia is a common complication. Regular



hemoglobin testing helps healthcare providers adjust treatment plans and evaluate the effectiveness of interventions aimed at improving hemoglobin levels and, consequently, overall patient health. Understanding the structure, function, and clinical significance of hemoglobin provides valuable insights into blood-related disorders and informs personalized treatment approaches[3]. Hemoglobin analysis remains a cornerstone in modern medicine, guiding healthcare practitioners in ensuring optimal oxygen delivery and overall patient well-being. The other one of the most common methods is spectrophotometry, which is based on the absorption of light by the iron atoms in the sample. This method allows for the determination of the iron concentration present in hemoglobin. By separating components based on their chemical properties, chromatography offers a detailed examination of the iron content. High-resolution chromatography is particularly useful for detecting even trace amounts of iron in hemoglobin samples[2].

Another method used is chromatography, which separates the components of a sample based on their chemical properties. High-resolution chromatography is especially useful for analyzing iron in hemoglobin due to its ability to detect trace amounts of substances. Mass spectrometry is a sensitive technique widely used for identifying and quantifying individual atoms, such as iron in hemoglobin. Its high accuracy and sensitivity make it a powerful tool in researching iron metabolism disorders and understanding the intricate details of iron levels within hemoglobin[2,3].

Furthermore, mass spectrometry is a sensitive and precise technique used to identify and quantify individual atoms, such as iron in hemoglobin. This technique has proven to be very valuable in researching disorders related to iron metabolism[4].

In summary, the analysis of iron content in hemoglobin is crucial for better understanding the function of this protein in the human body, as well as for diagnosing and treating diseases related to iron deficiency. The combination of different analytical techniques, such as spectrophotometry, chromatography, and mass spectrometry, provides a comprehensive view of the iron concentration in hemoglobin and its clinical relevance. The analysis of iron content in hemoglobin is essential for medical diagnostics and research. Utilizing a combination of these analytical methods - spectrophotometry, chromatography, and mass spectrometry - offers a comprehensive approach to studying the iron levels in hemoglobin and their clinical implications. A thorough understanding of the iron content in hemoglobin aids in diagnosing conditions associated with iron deficiency, contributing to improved patient care and health outcomes.

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